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11-2 Practice Problems. Stoichiometry and Balancing Equations. For each reaction balance the equation first. Determine the mass of lithium hydroxide, LiOH, produced when 0.38 g of lithium nitride, Li₃N, reacts with water according to the following equation: Li₃N + 3H₂O (NH₃ + 3LiOH.

~~11-2 Practice Problems - Brainly~~

Honors Chemistry 11-2 Practice problems Name: _____ See sample problem 2 on pg. 357 for the following problems: 1. Determine the mass of zinc chloride produced when 5.30 grams of zinc react with hydrochloric acid. 2. Determine the mass of lithium hydroxide produced when 0.38g of lithium nitride reacts with water according the the following ...

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11-2 Practice Problems Continued Chemistry Answers

Name: _____ Date: _____ Chemistry Chapter 11 Prentice-Hall, Inc. 13 11-2 Practice Problems 1. Determine the mass of Lithium Hydroxide produced when 0.38-g of Lithium Nitride reacts with water according to the following equation: $\text{Li}_3\text{N} + 3\text{H}_2\text{O} \rightarrow \text{NH}_3 + 3\text{LiOH}$

[k_g_k_sch_cmljaGFyZHNvbmpAbmV3dG9uLmsxMi5pYS51ew_9...](#)

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11-2 Practice Problems Chemistry Answers—Orris

Honors Chemistry 11-2 Practice problems Name: _____ See sample Page 10/22. Read PDF 11 2 Practice Problems Continued Answers problem 2 on pg. 357 for the following problems: 1. Determine the mass of zinc chloride produced when 5.30 grams of zinc react with hydrochloric acid. 2.

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330 Chapter 11 Print •Guided Reading and Study Workbook, Section 11.2 •Core Teaching Resources, Section 11.2 Review •Laboratory Manual, Labs 14, 15 •Small-Scale Chemistry Laboratory Manual, Lab 15 •Transparencies, T118–T120 •Lab Practical 11–1 Technology •Interactive Textbook with ChemASAP, Simulation 12; Problem-Solving 11.14, 11.15, 11.17, 11.18, 11.21; Assessment

11.2 Types of Chemical Reactions

Section 11.2 Stoichiometric Calculations pages 373–378 Practice Problems pages 375–377 11. Methane and sulfur react to produce carbon disulfide (CS_2), a liquid often used in the production of cellophane. $\text{CH}_4(\text{g}) + \text{S}_8(\text{s}) \rightarrow \text{CS}_2(\text{l}) + \text{H}_2\text{S}(\text{g})$ a. Balance the equation. $2\text{CH}_4(\text{g}) + \text{S}_8(\text{s}) \rightarrow 2\text{CS}_2(\text{l}) + 4\text{H}_2\text{S}(\text{g})$ b. Calculate the moles of CS_2

Stoichiometry Stoichiometry

Choose problems at the end of the chapter and take the practice exam in an exam like environment. Time it for the exam duration (e.g. 50 or 75 minutes), no open book or notes, just the equation sheet and periodic table included with the exam and the calculator that you will be using for the exam.

General Chemistry II

Description: 11-2 Practice Problems . 1. Determine the mass of lithium hydroxide ... decomposes explosively into carbon, carbon monoxide, hydrogen, and nitrogen.

Name [11-2 Practice Problems - Harrison High School Pages 1...](#)

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Name (please print): _____ Practice Exam 2 Chemistry 142, Spring 2005 Do only Problems 1,3,4,7,8, and 11 for in preparation for Test 2, 2016 Part 1. Answer 7 of the following 8 multiple choice questions. If you answer more than 8 cross out the one you wish not to be graded, otherwise only the first 8 will be graded. 4 points each.

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