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by ...

It consists of one carbon atom ( $A_r = 12$ ) and two oxygen atoms ( $A_r = 16$ ):  $M_r$  of  $CO_2 = 12 + 16 + 16 = 44$ . It could also be calculated this way:  $M_r$  of  $CO_2 = (1 \times 12) + (2 \times 16) = 12 + 32 = 44$ . Like ...

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