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Pure Substances and Mixtures!
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Chemical Changes

Understanding Atoms, elements, and
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Layer, atmosphere, ozone formation,
chlorofluorocarbons, CFC's 1.2
Chemistry and Matter Chemistry
Central Science

Chapter Summaries – Chemistry Matter and Change

Chapter 1 Section 2 MATTER and Its

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Properties Matter – Anything that has MASS and VOLUME. ATOMS and MOLECULES the basic building blocks of matter. ATOM-- the smallest unit of an element that still maintains the properties that element.
ELEMENT– a pure substance made up of only one type of atom.

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**MATTER and CHANGE Chapter 1
Section 1**

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Chapter 6 Study Guide for Content
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MASTERY Section 6.3 Periodic

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Trends In your textbook, read about atomic radius and ionic radius. Circle the letter of the choice that best completes the statement or answers the question. 1. Atomic radii cannot be measured ...

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CHAPTER 3 SOLUTIONS MANUAL

Matter—Properties and

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Section 3.1 Properties of Matter pages
70–75 Problem-Solving Lab 1. Explain
why the flow of a compressed gas
must be controlled for practical and
safe use.

**Matter—Properties and
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Chapter 1: Introduction to Chemistry

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Guide - Chapter 6 – The Periodic

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Table and Periodic Law Section 6.1
Development of the Modern Periodic
Table 1. octaves 2. eight 3. nine 4.
accepted 5. Dmitri Mendeleev 6.
atomic mass 7. elements 8. atomic
number 9. Henry Moseley 10. protons
11. periodic law 12. properties 13. 14
...

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Ch 6 Study Guide answers

Date Class Chapter Assessment

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Chapter 5 25 Electrons in Atoms

Reviewing Vocabulary Match the

definition in Column A with the term in

Column B. Column A Column B 1. The

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set of frequencies of the electromagnetic waves emitted by the atoms of an element 2. The minimum amount of energy that can be lost or gained by ...

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Change. extensive property. intensive

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property. extensive properties
(examples) intensive property
(examples) a property that depends on
the amount of matter in a sample. a
property that depends on the type of
matter in a sample not.... mass,
volume, length, weight.

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Chapter 11 31 number of moles of
known substance C H C H C H O O O
CO CO CO H O H O H O number of
moles of unknown substance O CO H

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OCHCOHOCHOHOCHO CO

use mole ratio 5 mol O 2 1 mol C 3 H

8 3 mol CO 2 1 mol C 3 H 8 4 mol H 2

O 1 mol C 3 H 8 1 mol C 3 H 8 5 mol

O 2

MATH SKILLS TRANSPARENCY
MASTER 16

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A “chemical change” is a change that produces matter with a different composition than the original matter. Chemical Change A change in which one or more substances are converted into different substances. Heat and light are often evidence of a chemical change. Properties of Compounds

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Quite different properties than their
component elements.

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