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Electrons move in circular orbits . around .  
the nucleus . at . fixed. energy. levels.

Electrons are never between energy levels  
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just the right amount of energy. to jump from one level to another. A . ... Chapter 13 Electrons in Atoms Last modified by:

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PLAY.  $3 \times 10^8$  m/s. What is the speed of

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light? meters. What unit must wavelength be in?  $E=h\nu$ . Equation to solve for energy.  $\nu = E/h$ . Equation for frequency. Joules. In what units is energy measured.  $6.626 \times 10^{-34} \text{ J}\cdot\text{s}$ . Planck's constant. Speed of light formula.  $c = \lambda \times \nu$ .

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we can "see" Microwaves, radiowaves, x-rays, ultraviolet radiation, infrared radiation -the forms of electromagnetic radiation that we can't see.

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Light and You will express the arrangements of electrons in atoms through orbital notations, electron

Analysis Answers will vary. Students will determine that observations typically rely heavily upon sight, ...

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chapter 5 electrons in atoms 138 Chapter 5

□ Electrons in Atoms Although the speed of all electromagnetic waves in a vacuum is the same, waves can have different wavelengths and frequencies. As you can see from the equation on the previous page, wavelength and frequency are inversely related; in other words, as one

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quantity increases, the ...

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