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Question 2. a) A is
an organic

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Compound which
molecular formula
is $C_4H_{10}O_2$. A
was gone under
some tests and
observations were
noted and listed as
below. a gas was
emitted when
sodium was added
to the A. A does not
react with with
NaOH. There are
two optical

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isomerism carbon
atoms on A.

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carbon and
hydrogen. Organic
compounds which
contain only carbon
and hydrogen are

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defined as
Hydrocarbons.
Many organic
compounds contain
nitrogen, oxygen,
halogens, sulfur,
phosphorus which
are derived from
replacing one or
more hydrogen
atoms in their
molecules with
these atoms.

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a) Design chemical
products and
process that
maximize profits b)

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Design safer
chemical products
and processes that
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products and
processes that
work most
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Chemistry quizzes
on organic
chemistry. Over
ninety percent of all
chemical
compounds are
classified as organic
compounds. They
were given this
name because
scientists once
thought that they
could only be made
naturally - not

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unnaturally by
scientists in a
laboratory.

However, in 1828,
Friedrich Wohler
made the naturally
occurring organic
compound urea in
his laboratory
which opened the
door to making
others from simpler
materials.

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Compounds
Title: Organic
chemistry in high
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Ichiro Ochiai
Subject: Journal of
Chemical Education,
Vol. 65 No.7, July
1988 p604,
Provocative Opinion

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assignment due
today and I don't
know which one to
choose. ...

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