

## October November 2012 Chemistry Paper 62 0620

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0620 w12 ms 13 - Papers

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0620 w12 ms 62 - Past Papers | GCE Guide

Page 3 Mark Scheme Syllabus Paper IGCSE October/November 2012 0620 33. © Cambridge International Examinations 2012. 3 (a) explanation of evaporation e.g. particles (or molecules) with a lot of energy leave the liquid / bromine particles break free from each other / forces or bonds between bromine molecules broken / molecules (in liquid) have weak forces holding them together / weak intermolecular forces / Van der Waals forces between molecules (don't have to be stated as weak) / (weak ...

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CHEMISTRY 5070/12 Paper 1 Multiple Choice October/November 2012 1 hour Additional Materials: Multiple Choice Answer Sheet Soft clean eraser Soft pencil (type B or HB recommended) READ THESE INSTRUCTIONS FIRST Write in soft pencil. Do not use staples, paper clips, highlighters, glue or correction fluid.

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### 0620 w12 ms 31 - Revision Science

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### IGCSE Chemistry 2012 Past Papers - CIE Notes

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### 5070 w12 ms 22 - O'Level Past Papers

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17/1/2017: October/November 2017 IGCSE Chemistry Grade Thresholds, Syllabus and Past Exam Papers are updated. 16/08/2018 : IGCSE Chemistry 2018 Past Papers of March and May are updated. 18 January 2019 : October / November 2018 papers are updated. Feb / March and May / June 2019 papers will be updated after result announcements.

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